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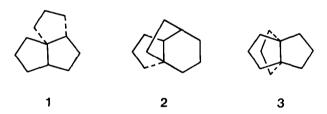
SYNTHESIS AND REARRANGEMENT OF FUNCTIONALIZED DISPIRO[3.0.3.3]UNDECANES A NEW ENTRY TO [3.3.3]PROPELLANES 1)

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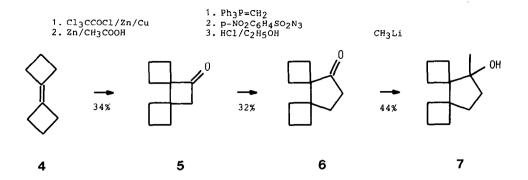
Summary: The functionalized dispiro[3.0.3.3]undecanes $\underline{6}$ and $\underline{7}$ undergo cascade rearrangements to yield the [3.3.3]propellanes $\underline{15}$ and $\underline{11}$, respectively. The formation of $\underline{15}$ proceeds via the bicyclic enone $\underline{16}$.

Naturally occurring sesquiterpenes based on the tricycloundecane skeletons $\underline{1}$, $\underline{2}$ and $\underline{3}$ have been the focus of considerable synthetic interest during the past few years 2). Molecular mechanics calculations 3) predict $\underline{1}$ ($\Delta H_f^o = -26.7 \, \text{kcal/mol}$) and $\underline{2}$ ($\Delta H_f^o = -25.7 \, \text{kcal/mol}$) to be thermodynamically favoured over the vaste majority of their tricycloundecane congeners, but $\underline{3}$ ($\Delta H_f^o = -29.6 \, \text{kcal/mol}$) is predicted to be the most stable of all. It therefore seemed particularly attractive to induce cascade rearrangements in suitable sized dispiranes in order to enter the tricycloundecane energy surface specifically near $\underline{1}$ and to look whether derivatives of $\underline{1}$, $\underline{2}$ and/or $\underline{3}$ would be obtained.



Out of several promising candidates 4 , we chose the dispiranes $\underline{6}$ and $\underline{7}$ as best suited for an initial rearrangement to the skeleton of $\underline{1}^{5}$. $\underline{6}$ and $\underline{7}$ were obtained as follows: addition of dichloroketene 6) to bicyclobutylidene $\underline{4}^{7}$ and subsequent dechlorination 6) of the resulting dichloroketone gave the dispiroketone $\underline{5}^{8}$) which was then homologated to $\underline{6}^{8}$) by a sequence of methylenation, reaction with p-nitrobenzenesulfonic acid azide and hydrolysis of the resulting ring expanded imide 9). Addition of methyllithium then yielded $\underline{7}^{8}$, albeit extensive enolization led to the recovery of up to 50% of unchanged $\underline{6}$.

When alcohol $\underline{7}$ was treated with an equimolar amount of a 0.54 molar solution of anhydrous p-toluenesulfonic acid in benzene for 2 h at $+70^{\circ}$ C, a quantitative conversion to the propellane $\underline{11}^{8}$) was observed. However, treatment of ketone $\underline{6}$ with



an equimolar amount of the same solution for 12 h at $+20^{\circ}$ C resulted in a quantitative conversion to the bicyclic enone $\underline{16}^{8}$. The same conversion was complete within 10 min at $+70^{\circ}$ C, but after 12 h at $+70^{\circ}$ C, the propellanone $\underline{15}^{10}$ had formed instead.

These results may be rationalized as follows: after protonation $(\underline{6},\underline{7})$ and dehydration $(\underline{7})$ both $\underline{6}$ and $\underline{7}$ undergo the expected $\underline{5}$) twofold cyclobutylmethyl-cyclopentyl rearrangement to yield the tricycloundecyl carbenium ions $\underline{8}$ and $\underline{12}$, respectively. At this stage a rapid but reversible ring opening of the $\underline{6}$ -hydroxycarbenium ion

$$7 = \begin{array}{c} +H^{+} \\ -H_{2}O \\ -H^{+} \end{array}$$

$$8 = \begin{array}{c} +H^{+} \\ +H^{+} \end{array}$$

$$10 = \begin{array}{c} -H^{+} \\ +H^{+} \end{array}$$

$$6 \stackrel{+H^{+}}{\rightleftharpoons} \stackrel{-H^{+}}{\rightleftharpoons} \stackrel{OH}{\rightleftharpoons} \stackrel{OH}{\rightleftharpoons}$$

12 to the bicyclic enone 16 occurs, while 8 undergoes two further 1,2-shifts to yield the propellane 11 (8-9-10-11). To account for the final conversion of the bicyclic enone 16 to the propellanone 15 a carbenium ion mediated transannular ring closure to 12 13) followed by two 1,2-shifts (16-12-13-14) may be anticipated. Support comes from the fact that 15 ($\Delta H_f^0 = -54.0 \text{ kcal/mol}$) is predicted to be thermodynamically favoured over both 16 ($\Delta H_f^0 = -40.7 \text{ kcal/mol}$) and 6 ($\Delta H_f^0 = -5.5 \text{ kcal/mol}$).

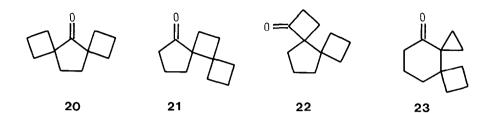
Finally, the ring opening of $\underline{12}$ leading to the bicyclic enone $\underline{16}$ deserves comment. This opening is an intramolecular variant of the known fragmentation of the 1,3-diol $\underline{17}$ which also proceeds via a β -hydroxycarbenium ion ($\underline{18}$) yielding tetramethylethylene $\underline{19}$ and acetone. The potential value of this fragmentation for the synthesis of other acyclic, mono- and bicyclic enones seems obvious.

In summary, the synthesis and rearrangement of suitable functionalized dispiro[3.0.3.3]undecanes provides a new and efficient entry to tricycloundecanes. Acid catalyzed rearrangements in nonnucleophilic solvents have shown to give derivatives of $\underline{3}$, acid catalyzed rearrangements in nucleophilic solvents and rearrangements induced by thionylchloride in pyridine should lead to derivatives of $\underline{1}$ and/or $\underline{2}$. The synthesis of naturally occurring sesquiterpenes based on the skeletons of $\underline{1}$, $\underline{2}$ and $\underline{3}$ might thus be feasible. Research towards this goal is in progress.

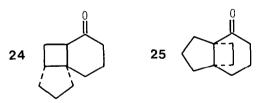
<u>Acknowledgement</u>: Financial support of the Fonds der Chemischen Industrie is gratefully acknowledged.

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- 8) All new compounds gave correct elemental analyses and/or high resolution mass spectral data. IR, $^{1}\mathrm{H}$ NMR, $^{13}\mathrm{C}$ NMR and mass spectral data are in accord with the structures given. $^{13}\mathrm{C}$ NMR data (20 MHz, CDCl3, unless otherwise stated): 5: 15.68, 16.16, 26.00, 29.21, 39.51, 55.93, 66.92, 212.65; 6: 14.33, 14.94, 23.90, 26.46, 30.58, 33.66, 47.89, 56.77, 219.82; 7: 15.02, $^{15.96}$, 21.53, 24.16 (multiplet selection revealed the coincidence of a primary and a secondary carbon atom), 29.90, 31.09, 34.44, 36.83, 51.20, 55.87, 81.93; 11: 13.82, 25.98, 37.75, 41.85, 47.48, 60.30, 70.17, 122.12, 144.07; 16 (C6D6): 22.26, 23.17, 24.75, 28.24, 36.12, 37.28, 40.43, 45.85, 136.51, $^{137.21}$, 211.50.
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